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- [54] Imidazoisoquinoline compounds and processes for preparation thereof.
- 57 Imadazoisoquinoline compounds of the formula:

and its preparation.

wherein

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R1 is lower alkyl,

R3 is hydrogen, halogen or ar(lower)alkoxy, and

R is lower alkanoyl, nitroso, amino, carboxy, protected carboxy, carbamoyl, hydroxycarbamoyl, haloformyl, aminomethyleneamino which may be substituted with cyano or lower alkyl, iminomethylamino which may be substituted with cyano or lower alkyl, or a group of the formula: $-A-R^2$ in which A is lower alkylene and

R² defines certain substituents, pharmaceutical salts thereof and pharmaceutical compositions containing them as an active ingredient. The invention also relates to intermediates of formula

IMIDAZOISOQUINOLINE COMPOUNDS AND PROCESSES FOR PREPARATION THEREOF

The present invention relates to novel imidazoisoquinoline compounds and pharmaceutically acceptable salt thereof. More particularly, it relates to novel imidazoisoquinoline compounds and pharmaceutically acceptable salts thereof which have inhibitory activity on ulcer, to processes for preparation thereof, to pharmaceutical composition comprising the same, and to method of using the same therapeutically in the treatment of ulcer in human being and animals.

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Accordingly, one object of this invention is to provide novel imidazoisoquinoline compounds and pharmaceutically acceptable salt thereof, which are useful as a medicine for ulcer.

Another object of this invention is to provide processes for preparation of said imidazoisoquinoline compounds and pharmaceutically acceptable salts thereof.

A further object of this invention is to provide

pharmaceutical composition comprising, as an active ingredient, said imidazoisoquinoline compound or its pharmaceutically acceptable salt.

Still further object of this invention is to provide method of using said imidazoisoquinoline compound or its pharmaceutically acceptable salt in the treatment of ulcer in human being and animals.

The imidazoisoquinoline compounds of this invention are novel and can be represented by the following general formula (I):

$$\mathbb{R}^{3} \longrightarrow \mathbb{R}^{1}$$
 (I)

wherein R is lower alkyl,

R³ is hydrogen, halogen or ar(lower)alkoxy, and

is lower alkanoyl, nitroso, amino, carboxy, protected carboxy, carbamoyl, hydroxy-carbamoyl, haloformyl, aminomethyleneamino which may be substituted with cyano or lower alkyl, iminomethylamino which may be substituted with cyano or lower alkyl, or a group of the formula: -A-R² in which A is lower alkylene and

R² is di(lower)alkylamino, cyano, lower alkoxy, N-containing heterocyclic group which may have suitable substituent(s), lower alkynyloxy, lower alkenyloxy, lower alkylthio,

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amino(lower) alkylthio, lower alkylsulfinyl, lower alkylsulfinyl, lower alkylsulfonyl, carboxy, protected carboxy, carbamoyl, hydroxy, hydroxycarbamoyl, hydroxy, lower alkanoyloxy, heterocyclicamino(lower) alkylthio having two oxo groups, hydrogen or a group of the formula:

 $X \Theta$ $\bigoplus_{X} \mathbb{R}^4 \quad \text{in which } \mathbb{R}^4, \mathbb{R}^5 \text{ and}$ $= \mathbb{R}^5 \quad \mathbb{R}^6 \quad \text{are each lower}$ $= \mathbb{R}^6 \quad \text{alkyl and}$ $X \quad \text{is an acid residue.}$

According to this invention, the object compounds (I) and their salts can be prepared by the following processes.

Process 1

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$$R^{3} \xrightarrow{H-R_{a}^{2} \text{ (III) or a salt thereof}} \mathbb{R}^{1} \xrightarrow{\mathbb{R}^{1}} \mathbb{R}^{1}$$
(II) (Ia)

or a salt thereof

or a salt thereof

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Process 3

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or a salt thereof

Process 4

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$$R^{3} \xrightarrow{N} R^{1}$$

$$R^{3} \xrightarrow{N} R^{1}$$
or a salt thereof

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$$\mathbb{R}^3$$
 Formylation \mathbb{R}^3 \mathbb{R}^1 (Ig)

or a salt thereof

or a salt thereof

Process 6

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Process 7

or a salt thereof

or a salt thereof

Process 9

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$$R^3$$
 R^3
 R^3

Process 10

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or a salt thereof

Process 11

Process 12

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or a salt thereof

or a salt thereof or a salt thereof

or a salt thereof

Process 14

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30 (It)

or a salt thereof

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or a salt thereof

or a salt thereof

Process 16

$$R^3$$
 + R^{12} -COOH (XII)

(Iw)
or a salt thereof

or its reactive derivative at the carboxy group or a salt thereof

(Ix)

or a salt thereof

or a salt thereof

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or a salt thereof

•				
20	wherein	R,	R ^I ,	R^3 , R^4 , R^5 , R^6 , A and X are each as
		_		defined above,
•		R_a^2	is	di(lower)alkylamino or N-containing
_				heterocyclic group which may have
		_		suitable substituent(s),
25		R _b ²	is	cyano, lower alkoxy, di(lower)alkylamino,
				N-containing heterocyclic group which may
				have suitable substituent(s), lower
				alkynyloxy, lower alkenyloxy, lower
				alkylthio, amino(lower)alkylthio or
30				heterocyclicamino(lower)alkylthio having
		_		two oxo groups,
		$R_{\rm C}^2$	is	tetrazol-5-yl,
		_R 7	is	lower alkyl,

methylamino,

R_a is cyanoaminomethyleneamino or cyanoimino-

R⁸ is lower alkyl, R_h is lower alkylaminomethyleneamino or lower alkyliminomethylamino, A^l is lower alkylene, R is heterocyclic group having two oxo groups, Y^l is halogen, R_s^2 is lower alkylthio, R_{a}^{2} is lower alkylsulfinyl or lower alkylsulfonyl, R¹⁰ is lower alkyl, $\mathbf{R}_{\mathbf{C}}$ is protected carboxy or protected carboxy(lower)alkyl, R_A is carboxy or carboxy(lower)alkyl, R¹¹ is hydrogen or hydroxy, R is carbamoyl, hydroxycarbamoyl, carbamoyl-(lower)alkyl or hydroxycarbamoyl(lower)alkyl, R_f is carboxy or esterified carboxy, \mathbb{R}^{12} is lower alkyl, and Y² is halogen.

Some of the starting compounds (II) are novel and can be prepared by the following Process.

Process A

$$R_{a}^{3} \longrightarrow NH_{2} + R_{-C-CH_{2}-Y}^{0}$$
(XIIIa)

or a salt thereof

$$\mathbb{R}^{3} = \mathbb{R}^{1}$$
(IIa)

or a salt thereof

wherein R¹ is as defined above,
R³ is halogen or ar(lower)alkoxy, and
Y³ is halogen.

Suitable pharmaceutically acceptable salts of the object compounds (I) are conventional non-toxic salts and include a metal salt such as an alkali metal salt (e.g. sodium salt, potassium salt, etc.) and an alkaline earth metal salt (e.g. calcium salt, magnesium salt, etc.), an ammonium salt, an organic base salt (e.g. trimethylamine salt, triethylamine salt, pyridine salt, picoline salt, dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, etc.), an organic acid

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salt (e.g. acetate, maleate, tartrate, methanesulfonate, benzenesulfonate, formate, toluenesulfonate, etc.), an inorganic acid salt (e.g. hydrochloride, hydrobromide, sulfate, phosphate, etc.), a salt with an amino acid (e.g., arginine, aspartic acid, glutamic acid, etc.), and the like.

In the above and subsequent descriptions of the present specification, suitable examples and illustrations of the various definitions which the present invention include within the scope thereof are explained in detail as follows.

The term "lower" is intended to mean I to 6 carbon atom(s), unless otherwise indicated.

Suitable "lower alkyl" and "lower alkyl moiety" in the terms "aminomethyleneamino which may be substituted with cyano or lower alkyl", "iminomethylamino which may be substituted with cyano or lower alkyl", "di(lower) alkylamino", "lower alkylthio", "amino(lower) - alkylthio", "lower alkylsulfinyl", "lower alkylsulfonyl", "heterocyclicamino(lower) alkylthio having two oxo groups", "lower alkylaminomethyleneamino", "lower alkyliminomethylamino", "carboxy(lower)alkyl", "protected

carboxy(lower)alkyl", "carbamoyl(lower)alkyl" and "hydroxycarbamoyl(lower)alkyl" may include methyl, ethyl, propyl, isopropyl, butyl, t-butyl, pentyl, hexyl and the like.

Suitable "halogen" may include chlorine, bromine, fluorine, or iodine.

Suitable "ar(lower)alkoxy" may include phenyl(lower)-alkoxy (e.g., benzyloxy, phenethyloxy, etc.) and the like.

Suitable "lower alkanoyl" and "lower alkanoyl moiety" in the term "lower alkanoyloxy" may include formyl, acetyl, propionyl, butyryl, isobutyryl, valeryl, isovaleryl, hexanoyl and the like.

Suitable "protected carboxy" and "protected carboxy moiety" in the term "protected carboxy(lower)-alkyl" may include esterified carboxy and the like.

Suitable "ester moiety" in the term "esterified carboxy" may be the ones such as lower alkyl ester (e.g. methyl ester, ethyl ester, propyl ester, isopropyl ester, butyl ester, isobutyl ester, tert-butyl ester, pentyl ester, hexyl ester, 1-cyclopropylethyl ester, etc.) which may have at least one suitable substituent(s), for example, lower alkanoyloxy(lower)alkyl ester [e.g. acetoxymethyl ester, propionyloxymethyl ester, butyryloxymethyl ester, valeryloxymethyl ester, pivaloyloxymethyl ester, hexanoyloxymethyl ester, 1(or 2)-acetoxyethyl ester, 1(or 2 or 3)-acetoxypropyl ester, 1(or 2 or 3 or 4)-acetoxybutyl ester, 1(or 2)propionyloxyethyl ester, 1(or 2 or 3)-propionyloxypropyl ester, 1(or 2)-butyryloxyethyl ester, 1(or 2)isobutyryloxyethyl ester, 1(or 2)-pivaloyloxyethyl ester, 1 (or 2) -hexanoyloxyethyl ester, isobutyryloxymethyl ester, 2-ethylbutyryloxymethyl ester, 3,3-dimethylbutyryloxymethyl ester, 1(or 2)-pentanoyloxyethyl ester, etc.], lower alkanesulfonyl (lower) alkyl ester (e.g. 2-mesylethyl

ester, etc.), mono(or di or tri)-halo(lower)alkyl ester (e.g. 2-iodoethyl ester, 2,2,2-trichloroethyl ester, etc.), lower alkoxycarbonyloxy(lower)alkyl ester (e.g. methoxycarbonyloxymethyl ester, ethoxycarbonyloxymethyl 5 ester, 2-methoxycarbonyloxyethyl ester, 1-ethoxycarbonyloxyethyl ester, 1-isopropoxycarbonyloxyethyl ester, etc.), or phthalidylidene(lower)alkyl ester; lower alkenyl ester (e.g. vinyl ester, allyl ester, etc.); lower alkynyl ester (e.g. etynyl ester, propynyl ester, 10 etc.); ar(lower)alkyl ester [e.g., mono(or di or tri)phenyl-(lower) alkyl ester, etc.] which may have at least one suitable substituent(s) (e.g. benzyl ester, 4methoxybenzyl ester, 4-nitrobenzyl ester, phenethyl 15 ester, trityl ester, benzhydryl ester, bis (methoxyphenyl) methyl ester, 3,4-dimethoxybenzyl ester, 4-hydroxy-3,5-di-tert-butylbenzyl ester, etc.); aryl ester which may have at least one suitable substituent(s) (e.g. phenyl ester, 4-chlorophenyl 20 ester, tolyl ester, tert-butylphenyl ester, xylyl ester, mesityl ester, cumenyl ester, etc.), and the like.

Suitable "haloformyl" may include fluoroformyl, chloroform, bromoformyl and iodoformyl.

Suitable "lower alkylene" may include methylene, ethylene, trimethylene, tetramethylene, pentamethylene or hexamethylene.

Suitable "lower alkoxy" may include methoxy, ethoxy, propoxy, isopropoxy, butoxy, t-butoxy pentyloxy, hexyloxy and the like.

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Suitable "N-containing heterocyclic group moiety" in the term "N-containing heterocyclic group which may have suitable substituent(s)" means saturated or unsaturated, monocyclic or polycyclic heterocyclic group containing nitrogen atom(s). And, especially

preferable N-containing heterocyclic group may be heterocyclic group such as

unsaturated 3 to 8-membered heteromonocyclic group containing 1 to 4 nitrogen atom(s), for example, pyrrolyl, pyrrolinyl, imidazolyl, pyrazolyl, triazolyl (e.g., 4H-1,2,4-triazolyl, 1H-1,2,3-triazolyl, 2H-1,2,3-triazolyl, etc.), tetrazolyl (e.g., 1H-tetrazolyl, 2H-tetrazolyl, etc.), dihydrotriazinyl, etc.; saturated 3 to 8-membered heteromonocyclic group containing 1 to 4 nitrogen atom(s), for example, pyrrolidinyl, imidazolidinyl, piperidino, piperazinyl, etc.; unsaturated condensed heterocyclic group containing 1 to 5 nitrogen atom(s), for example, indolyl, isoindolyl, benzimidazolyl, 1H-indazolyl, etc.

Suitable substituent(s) in the term "N-containing heterocyclic group which may have suitable substituent(s)" may include lower alkyl (e.g., methyl, ethyl, propyl, isopropyl, butyl, isobutyl, t-butyl, pentyl, hexyl, etc.), hydroxy(lower)alkyl [e.g., hydroxymethyl, 1 or 2-hydroxyethyl, 1 or 2 or 3-hydroxypropyl, 1 or 2 or 3 or 4-hydroxybutyl, 1 or 2 or 3 or 4 or 5-hydroxypentyl, 1 or 2 or 3 or 4 or 5 or 6-hydroxyhexyl, etc.], and the like.

Suitable "lower alkynyloxy" may include ethynyloxy, propynyloxy, butynyloxy pentynyloxy, hexynyloxy and the like.

Suitable "lower alkenyloxy" may include vinyloxy, propenyloxy, butenyloxy, pentenyloxy, hexenyloxy and the like.

Suitable "heterocyclic moiety" in the terms

"heterocyclicamino(lower)alkylthio having two oxo groups"

and "heterocyclic group having two oxo groups"

may be one containing at least one hetero atom selected

from nitrogen, sulfur and oxygen atom, and may include

saturated or unsaturated, monocyclic or polycyclic

heterocyclic group, and preferable heterocyclic group. may be N-containing heterocyclic group such as unsaturated 3 to 6 membered heteromonocyclic group containing 1 to 4 nitrogen atoms, for example, pyrrolyl, pyrrolinyl, imidazolyl, pyrazolyl, pyridyl, pyrimidinyl, 5 pyrazinyl, pyridazinyl, triazolyl (e.g., 4H-1,2,4triazolyl, 1H-1,2,3-triazolyl, 2H-1,2,3-triazolyl, etc.), tetrazolyl (e.g., lH-tetrazolyl, 2H-tetrazolyl, etc.), etc.; 10 saturated 3 to 6-membered heteromonocyclic group containing 1 to 4 nitrogen atoms (e.g., pyrrolidinyl, imidazolidinyl, piperidino, piperazinyl, etc.); unsaturated condensed heterocyclic group containing 1 to 5 nitrogen atoms, for example, indolyl, isoindolyl, 15 indolizinyl, benzimidazolyl, quinolyl, isoquinolyl, indazolyl, benzotriazolyl, tetrazolopyridazinyl (e.g., tetrazolo[1,5-b]pyridazinyl, etc.), etc.; unsaturated 3- to 6-membered heteromonocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms, for example, oxazolyl, isoxazolyl, oxadiazolyl 20 (e.g., 1,2,4-oxadiazolyl, 1,3,4-oxadiazolyl, 1,2,5oxadiazolyl, etc.), etc.; saturated 3 to 6-membered heteromonocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms 25 (e.g., morpholinyl, etc.); unsaturated condensed heterocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms (e.g., benzoxazolyl, benzoxadiazolyl, etc.); unsaturated 3 to 6-membered heteromonocyclic group 30 containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms, for example, thiazolyl, thiadiazolyl (e.g., 1,2,4thiadiazolyl, 1,3,4-thiadiazolyl, 1,2,5-thiadiazolyl, etc.), etc.; saturated 3 to 6-membered heteromonocyclic group 35 containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms

(e.g., thiazolidinyl, etc.); unsaturated condensed heterocyclic group containing 1 to 2 sulfur atoms and 1 to 3 nitrogen atoms (e.g., benzothiazolyl, 1,2-benzisothiazolyl, 2,1-benzisothiazolyl, benzothiadiazolyl, etc.) and the like.

Suitable "acid residue" may include a residue of an acid such as hydrohalogenic acid [i.e. halogen (e.g., chlorine, bromine, fluorine or iodine)], (lower)-alkylsulfuric acid (e.g., methylsulfuric acid, ethylsulfuric acid, etc.) or the like.

Preferable embodiments of the object compounds (I) are as follows.

Preferable embodiment of R¹ is lower alkyl (more preferably methyl);

- R³ is hydrogen, halogen (more preferably chlorine)
 or ar(lower)alkoxy[more preferably phenyl(lower) alkoxy, most preferably benzyloxy]; and
- R is lower alkanoyl, nitroso, amino, carboxy, protected carboxy [more preferably esterified carboxy, most preferably lower alkoxycarbonyl], carbamoyl, hydroxycarbamoyl, haloformyl, cyanoaminomethyleneamino,

lower alkylaminomethyleneamino, cyanoiminomethylamino, lower alkyliminomethylamino, or a group of the formula : $-A-R^2$

in which A is lower alkylene (more preferably methylene) and

R² is di(lower) alkylamino (more preferably dimethylamino), cyano, lower alkoxy (more preferably methoxy or ethoxy),
N-containing heterocyclic group which may have lower alkyl or hydroxy(lower)-alkyl [more preferably saturated 3 to 8-membered heteromonocyclic group containing 1 to 4 nitrogen atom(s)

which have hydroxy(lower)alkyl or unsaturated 3 to 8-membered heteromonocyclic group containing 1 to 4 nitrogen atom(s) which may have lower alkyl, most preferably piperazinyl having hydroxy(lower)alkyl, tetrazolyl, imidazolyl or imidazolyl having lower alkyl], lower alkynyloxy, lower alkenyloxy, lower alkylthio, amino-(lower) alkylthio, lower alkylsulfinyl, lower alkylsulfonyl, carboxy, protected carboxy (more preferably esterified carboxy, most preferably lower alkoxycarbonyl), carbamoyl, hydroxycarbamoyl, hydroxy, lower alkanoyloxy, benzisothiazolylamino(lower)alkylthio having two oxo groups [more preferably (1,1-dioxo-1,2-benzisothiazoly1)amino(lower)alkylthio], hydrogen or a group of the formula :

 $\stackrel{\text{(+)}}{\sim} \underset{\mathbb{R}^5}{\overset{\mathbb{R}^4}{\sim}}$

in which R^4 , R^5 and R^6 are each lower alkyl,

X is an acid residue (more preferably residue of lower alkylsulfuric acid, or halogen).

The processes for preparing the object compounds of the present invention are explained in detail in the following.

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The compound (Ia) or a salt thereof can be prepared by reacting the compound (II) or a salt thereof with the compound (III) or a salt thereof and the compound (IV).

Suitable salts of the compounds (Ia) and (III) can be referred to the ones as exemplified for the compound (I).

Suitable salt of the compound (II) can be referred to the acid addition salt as exemplified for the compound (I).

The present reaction is usually carried out in the presence of a conventional acid. Suitable acid may include an organic acid (e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, etc.) and an inorganic acid (e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, etc.).

The reaction is usually carried out in a solvent such as water, methylene chloride, N,N-dimethyl-formamide, an alcohol (e.g. methanol, ethanol, etc.), a mixture thereof or any other solvent which does not adversely influence the reaction. A liquid acid can be also used as the solvent. The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Process 2

The compound (Ic) can be prepared by reacting the compound (Ib) or a salt thereof with the compound (V).

Suitable salt of the compound (Ib) can be referred to the acid addition salt as exemplified for the compound (I).

This reaction is usually carried out in a solvent such as alcohol (e.g. methanol, ethanol, etc.), benzene, N.N-dimethylformamide, tetrahydrofuran, acetone,

diethyl ether or any other solvent which does not adversely affect the reaction.

In case that the compound (V) to be used is liquid, it can also be used as a solvent.

The reaction temperature is not critical and the reaction is usually carried out under cooling to warming.

Process 3

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The compound (Id) or a salt thereof can be prepared by reacting the compound (Ic) with the compound (VI) or a salt thereof.

Suitable salts of the compounds (Id) and (VI) can be referred to the ones as exemplified for the compound (I).

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], dimethyl sulfoxide, benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether of any other solvent which does not adversely affect the reaction.

In case that the compound (VI) or a salt thereof to be used in liquid, it can also be used as a solvent.

The reaction temperature is not critical and the reaction is usually carried out at ambient temperature, or under warming or heating.

Process 4

The compound (If) or a salt thereof can be prepared by reacting the compound (Ie) or a salt thereof with an azide compound.

Suitable salts of the compounds (Ie) and (If) can be referred to the acid addition salt as exemplified for the compound (I).

Suitable example of the azide compound may be

an inorganic base salt of azide [e.g. sodium azide, potassium azide, lithium azide, calcium azide, barium azide, etc.], hydrogen azide, hydrazonic acid, ammonium azide or the like.

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under warming or heating.

Process 5

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The compound (Ig) or a salt thereof can be prepared by subjecting the compound (II) or a salt thereof to formylation reaction.

Suitable salt of the compound (Ig) can be referred to the acid addition salt as exemplified for the compound (I).

This formylation can be carried out by a conventional method, for example by reacting the compound (II) or a salt thereof with Vilsmeier reagent (prepared by the reaction of N,N-dimethylformamide with phosphoryl chloride, etc.) (first step) and then subjecting the resultant compound to hydrolysis (second step).

(i) First step:

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

.The reaction temperature is not critical and the

reaction is usually carried out under warming or heating.

(ii) Second step (hydrolysis) :

Hydrolysis can be carried out by a conventional method.

Process 6

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The compound (Ih) or a salt thereof can be prepared by subjecting the compound (II) or a salt thereof to nitrosation reaction.

Suitable salt of the compound (Ih) can be referred to the acid addition salt as exemplified for the compound (I).

Suitable nitrosating agent to be used in this reaction may include lower alkyl nitrite (e.g., t-butyl nitrite, amyl nitrite, isoamyl nitrite, etc.) and the like.

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, dioxane, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under warming or heating.

Process 7

The compound (Ii) or a salt thereof can be prepared by subjecting the compound (Ih) or a salt thereof to reduction reaction.

Suitable salt of the compound (Ii) can be referred to the acid addition salt as exemplified for the compound (I).

The reduction method applicable for this reaction may include, for example, reduction by using a combination of a metal (e.g. zinc, zinc amalgam, etc.)

or a salt of chrome compound (e.g. chromous chloride, chromous acetate, etc.) and an organic or inorganic acid (e.g. acetic acid, propionic acid, hydrochloric acid, etc.); and conventional catalytic reduction in the presence of a conventional metallic catalyst (e.g. palladium-carbon, etc.).

"This reaction is usually carried out in a solvent such as water, alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Process 8

The compound (Ij) or a salt thereof can be prepared by reacting the compound (Ii) or a salt thereof with the compound (VII).

Suitable salt of the compound (Ij) can be referred to the acid addition salt as exemplified for the compound (I).

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Process 9

The compound (Ik) or a salt thereof can be prepared by reacting the compound (Ij) or a salt thereof with the compound (VIII) or a salt thereof.

Suitable salts of the compounds (Ik) and (VIII) can

be referred to the acid addition salt as exemplified for the compound (I).

This reaction is usually carried out in a solvent such as water alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

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Process 10

The compound (Im) or a salt thereof can be prepared by reacting the compound (IL) or a salt thereof with the compound (IX) or a salt thereof.

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Suitable salts of the compounds (IL), (Im) and (IX) can be referred to the acid addition salt as exemplified for the compound (I).

This reaction is usually carried out in the presence of base.

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Suitable base may include an inorganic base such as alkali metal hydride (e.g. sodium hydride, etc.) alkali metal hydroxide (e.g. sodium hydroxide, potassium hydroxide, etc.), alkaline earth metal hydroxide (e.g. magnesium hydroxide, calcium hydroxide, etc.), alkali metal carbonate (e.g. sodium carbonate, potassium carbonate, etc.), alkaline earth metal carbonate (e.g. magnesium carbonate, calcium carbonate, etc.), alkali metal bicarbonate (e.g. sodium bicarbonate, potassium bicarbonate, etc.), alkali metal acetate (e.g. sodium acetate, potassium acetate, etc.), alkaline earth metal phosphate (e.g. magnesium phosphate, calcium phosphate, etc.), alkali metal hydrogen phosphate (e.g. disodium hydrogen phosphate, dipotassium hydrogen phosphate, etc.), or the like, and an organic base such as trialkylamine (e.g. trimethylamine, triethylamine,

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etc.), picoline, N-methylpyrrolidine, N-methylmorpholine or the like.

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Process 11

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The compound (Io) or a salt thereof can be prepared by subjecting the compound (In) or a salt thereof to oxidation reaction.

Suitable salts of the compounds (In) and (Io) can be referred to the acid addition salt as exemplified for the compound (I).

The present oxidation reaction can be carried out by a conventional method, for example by using an oxidizing agent such as m-chloroperbenzoic acid, perbenzoic acid, peracetic acid, ozone, hydrogen peroxide, periodic acid or the like.

This reaction is usually carried out in a solvent such as benzene, N,N-dimethylformamide, tetrahydrofuran, chloroform, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling or at ambient temperature.

Process 12

The compound (Iq) or a salt thereof can be prepared by reacting the compound (Ip) or a salt thereof with the compound (X) (first step) and then subjecting the resultant compound to hydrolysis (second step).

Suitable salts of the compounds (Ip) and (Iq) can be referred to the acid addition salt as exemplified for the compound (I).

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(i) First step:

This reaction is usually carried out in the presence of acid. Suitable acid may include an organic acid (e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, etc.) and an inorganic acid (e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, etc.).

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under warming or heating.

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(ii) Second step (hydrolysis) :

Hydrolysis can be carried out by a conventional method.

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Process 13

The compound (Is) or a salt thereof can be prepared by subjecting the compound (Ir) or a salt thereof to elimination reaction of the carboxy protective group.

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Suitable'salt of the compound (Is) can be referred to the ones as exemplified for the compound (I).

Suitable salt of the compound (Ir) can be referred to the acid addition salt as exemplified for the compound (I).

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The present-reaction is carried out in accordance

with a conventional method such as hydrolysis, reduction or the like.

Hydrolysis is preferably carried out in the presence of a base or an acid. Suitable base may include an inorganic base and an organic base such as an alkali metal (e.g. sodium, potassium, etc.), an alkaline earth metal (e.g. magnesium, calcium, etc.), the hydroxide or carbonate or bicarbonate thereof, trialkylamine (e.g. trimethylamine, triethylamine, etc.), or the like. Suitable acid may include an organic acid (e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, etc.) and an inorganic acid (e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, etc.).

Reduction is carried out in a conventional manner, including chemical reduction and catalytic reduction.

The reaction is usually carried out in a solvent such as water, methylene chloride, an alcohol (e.g. methanol, ethanol, etc.), a mixture thereof or any other solvent which does not adversely influence the reaction. A liquid base or acid can be also used as the solvent. The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Process 14

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The compound (It) or a salt thereof can be prepared by reacting the compound (Is) or its reactive derivative at the carboxy group or a salt thereof with the compound (XI) or its reactive derivative at the amino group or a salt thereof.

Suitable salts of the compounds (Is) and (It) can be referred to the ones as exemplified for the compound (I).

Suitable salt of the compound (XI) can be referred

to the acid addition salt as exemplified for the compound (I).

Suitable reactive derivative at the carboxy group of the compound (Is) may include an acid halide, an acid anhydride, an activated amide, an activated ester and the like.

This reaction is usually carried out in a solvent such as water alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to warming.

Process 15

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The compound (Iv) or a salt thereof can be prepared by subjecting the compound (Iu) or a salt thereof to reduction reaction.

Suitable salts of the compounds (Iu) and (Iv) can be referred to the ones as exemplified for the compound (I).

Suitable reducing agent may include lithium aluminum hydride and the like.

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling or at ambient temperature.

Process 16

The compound (Ix) or a salt thereof can be prepared

by reacting the compound (Iw) or a salt thereof with the compound (XII) or its reactive derivative at the carboxy group or a salt thereof.

Suitable salt of the compound (Iw) can be referred to the ones as exemplified for the compound (I).

Suitable salt of the compound (XII) may include an alkali metal salt (e.g., sodium salt, potassium salt, etc.) and the like.

Suitable salt of the compound (Ix) can be referred to the acid addition salt as exemplified for the compound (I).

Suitable reactive derivative at the carboxy group of the compound (XII) can be referred to the ones as exemplified for the compound (Is).

This reaction is usually carried out in a solvent such as benzene, N,N-dimethylformamide, tetrahydrofuran, pyridine, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to warming.

Process 17

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The compound (I) or a salt thereof can be prepared by reacting the compound (XIII) or a salt thereof with the compound (XIV) or a salt thereof.

Suitable salt of the compound (XIV) can be referred to the ones as exemplified for the compound (I).

Suitable salt of the compound (XIII) can be referred to the acid addition salt as exemplified for the compound (I).

This reaction is usually carried out in a solvent such as alcohol [e.g. methanol, ethanol, etc.], benzene, chloroform, N,N-dimethylformamide, tetrahydrofuran, diethyl ether or any other solvent which does not adversely affect the reaction.

The reaction may be preferably carried out in the presence of an inorganic or an organic base such as an alkali metal hydroxide [e.g. sodium hydroxide, potassium hydroxide, etc.], an alkali metal carbonate [e.g. sodium carbonate, potassium carbonate, etc.], an alkali metal bicarbonate [e.g. sodium bicarbonate, potassium bicarbonate, etc.], tri(lower) alkylamine [e.g. trimethylamine, triethylamine, etc.], pyridine or its derivative [e.g. picoline, lutidine, 4-dimethylaminopyridine, etc.], or the like. In case that the base to be used is liquid, it can also be used as a solvent.

The reaction temperature is not critical, and the reaction can be carried out under cooling, at ambient temperature or under warming or heating.

The process for preparing the starting compound (IIa) is explained in detail in the following.

30 Process A

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The compound (IIa) or a salt thereof can be prepared by reacting the compound (XIIIa) or a salt thereof with the compound (XV).

Suitable salts of the compounds (IIa) and (XIIIa) can be referred to the acid addition salt as exemplified for the compound (I). The present reaction can be carried out in a similar manner to that of aforementioned Process 17.

The object compounds (I) and their pharmaceutically acceptable salts of the present invention are novel and exhibit high inhibitory activity on ulcer.

In order to illustrate the usefulness of the object compound (I), the pharmacological data of some of the object compounds (I) are shown in the following.

(A) <u>Inhibition on ethanol ulcer</u> Test Method:

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Five male Sprague-Dawley rates, aged 7 weeks and weighing about 200 g, were used per group for the study on ethanol ulcer after the fast for 24 hours.

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Test compound was suspended in 0.1% methyl-cellulose aqueous solution, and the suspension (5 ml/kg) was orally given to each rat.

The control group was given a vehicle, i.e. 0.1% methylcellulose aqueous solution (5 ml/kg), alone in the same way.

Absolute ethanol (5 ml/kg) was orally administered 30 minutes after dosing with test compound, and one hour later, the rats were sacrificed and their stomachs were removed. The area of ulcers of each rat was measured. The mean area (mm²) in the medicated group was compared with that in the control group.

Test Result :

Test Compound: 7-Chloro-3-cyanomethyl-2-methylimidazo[2,1-a]isoquinoline
The ED₅₀ value of the test compound: 0.97 mg/kg

(B) <u>Inhibition on stress ulcer</u>

Test Method:

Five Sprague-Dawley rats weighing about 200 g were used per group. Each animal was immobilized in a small cage and put in a water bath allowing to respire. The temperature of the water bath kept at 22°C. The test compound was administered orally just before the immobilization. Seven hours later, the animals were sacrificed and their stomachs were removed. The stomach was then fixed with 2% formalin. The area of ulcers was measured for each animal. The mean area (mm²) in the medicated animals was compared with that in the control animals.

Test Result:

Test Compound : 7-Chloro-3-cyanomethyl-2- methylimidazo[2,l-a]isoquinoline The ${\rm ED}_{50}$ value of the test compound : 1.8 mg/kg

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As being apparent from the above test results, the object compound (I) of the present invention are useful as antiulcer medicines.

For therapeutic purpose, the compounds according to the present invention can be used in a form of pharmaceutical preparation containing said compound as an active ingredient, in admixture with a pharmaceutically acceptable carrier such as an organic or inorganic solid or liquid excipient suitable for oral or parenteral administration. The pharmaceutical preparations may be capsules, tablets, dragees, solution, suspension, emulsion, and the like. If desired, there may be included in the above preparations auxiliary substances, stabilizing agents, wetting or emulsifying agents, buffers and other commonly used additives.

While the dosage of the compounds will vary depending upon the age and condition of the patient, an average single dose of about 5 mg, 10 mg, 50 mg, 100 mg, 250 mg, 500 mg, and 1000 mg of the compounds according to the present invention may be effective for treating ulcer. In general, amounts between 1 mg/body and about 6,000 mg/body or even more may be administered per day.

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The following preparations and examples are given for the purpose of illustrating the present invention.

Preparation 1

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A mixture of 5-chloro-l-aminoisoquinoline (2.8 g), chloroacetone (5.4 g) and sodium bicarbonate (7.6 g) in anhydrous ethanol (35 ml) was heated at 60°C for 16 hours and filtered by suction. The filtrate was evaporated in vacuo and the residual solid was washed with ethanol to give 7-chloro-2-methylimidazo[2,1-a]isoquinoline (2.8 g).

mp : 188 to 191°C

IR (Nujol) : 1470, 1390, 800 cm⁻¹

NMR (CDCl₃, δ) : 2.47 (3H, s), 7.22-7.84 (5H, m), 8.40-8.53 (1H, m)

Preparation 2

The following compounds were obtained according to a similar manner to that of Preparation 1.

- (1) 7-Benzyloxy-2-methylimidazo[2,1-a]isoquinoline
 mp: 89 to 90°C
 IR (Nujol): 1602, 1550, 1510, 1482, 1260 cm⁻¹
 NMR (CDCl₃, δ): 2.50 (3H, s), 5.17 (2H, s), 6.93
 (1H, d, J=8Hz), 7.10-7.80 (9H, m), 8.14 (1H, d, J=8Hz)
- (2) 8-Chloro-2-methylimidazo[2,1-a]isoquinoline
 mp : 131 to 133°C
 IR (Nujol) : 1625, 1510 cm⁻¹
 NMR (DMSO-d₆, δ) : 2.38 (3H, s), 7.12 (1H, d, J=7.5Hz),
 7.58 (1H, dd, J=9.5Hz, 2Hz), 7.68 (1H, s), 7.92
 (1H, d, J=2Hz), 8.28 (1H, d, J=7.5Hz), 8.37
 (1H, d, J=9.5Hz)

Example 1

To a solution of 37% aqueous formaldehyde (0.42 g) and 50% aqueous dimethylamine (0.46 g) in acetic acid (3.2 ml) was added 7-chloro-2-methylimidazo[2,1-a]-isoquinoline (1 g) and the mixture was warmed at 50°C

with stirring for 3.5 hours. The solution was poured into ice water and neutralized with an aqueous solution of sodium hydroxide. The resulting precipitates were collected by filtration, washed with water and recrystallized from a mixture of toluene and n-hexane to give 7-chloro-3-dimethylaminomethyl-2methylimidazo[2,1-a]isoquinoline (0.9 g).

mp : 157 to 160°C

1520, 1480, 1400, 1380, 1370, IR (Nujol) : 1020, 780 cm⁻¹

NMR (CDCl₃, δ): 2.20 (6H, s), 2.47 (3H, s), 3.63 (2H, s), 7.24-8.60 (5H, m)

Analysis Calcd. for C15H16ClN3: C 65.81, H 5.89, N 15.35, Cl 12.95 Found: C 65.86, H 5.80, N 15.16, Cl 12.59

Example 2

The following compounds were obtained according to a similar manner to that of Example 1.

3-Dimethylaminomethyl-2-methylimidazo[2,l-a]isoquinoline

mp: 97 to 99°C

IR (Nujol): 1710, 1450, 1390, 1380, 1370 cm⁻¹ NMR (CDCl₃, δ): 2.24 (6H, s), 2.48 (3H, s), 3.65 (2H, s), 7.30 (1H, d, J=8Hz), 7.47-7.77

(3H, m), 8.06 (1H, d, J=8Hz), 8.64-8.73 (1H, m)

Analysis Calcd. for C15H17N3: C 75.28, H 7.16, N 17.56 Found: C 74.97, H 7.11, N 17.35

7-Benzyloxy-3-dimethylaminomethyl-2-methylimidazo-(2) [2,1-a]isoquinoline

135 to 137°C : qm IR (Nujol): 1600, 1560, 1510, 1480, 1263 cm⁻¹ NMR (CDCl₃, δ): 2.23 (6H, s), 2.50 (3H, s), 3.67 (2H, s), 5.28 (2H, s), 6.97-7.70 (8H, m), 8.00-8.40 (2H, m)

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8-Chloro-3-dimethylaminomethyl-2-methylimidazo[2,l-a]-(3) isoquinoline mp : 117 to 120°C IR (Nujol): 1635, 1515 cm⁻¹ NMR (DMSO- d_6 , δ): 2.18 (6H, s), 2.38 (3H, s), 5 3.70 (2H, s), 7.17 (1H, d, J=7.5Hz), 7.60 (1H, dd, J=9Hz, 2Hz), 7.95 (1H, d, J=2Hz), 8.22 (1H, d, J=7.5Hz), 8.38 (1H, d, J=9Hz) lO Example 3 To a solution of 37% aqueous formaldehyde (1.5 g), 1-(2-hydroxyethyl)piperazine (2.35 g), water (2 ml) and acetic acid (12 ml) was added 2-methylimidazo [2,1-a]isoquinoline (3 g) and the solution was warmed at 50°C with stirring for 3 hours. The solution was 15 poured into ice water and neutralized with an aqueous solution of sodium hydroxide. The resulting precipitates were collected by filtration, washed with water and chromatographed on 20 silica gel (60 g) with a mixture of chloroform and methanol (10:1) as an eluent. The eluates were evaporated in vacuo and the residual solid was recrystallized from a mixture of ethyl acetate and n-hexane to give 3-[4-(2-hydroxyethyl)-1-piperazinyl-25 methyl]-2-methylimidazo[2,l-a]isoquinoline (2.2 g). mp : 115 to 117°C IR (Nujol): 3200, 2800, 1515, 1410, 1340, 1160, 1080, 1010 cm⁻¹

NMR (CDCl₃, δ): 2.33-2.60 (13H, m), 3.63 (2H, s), 3.70 (2H, s), 7.00 (1H, d, J=8Hz), 7.33-7.78 (3H, m), 8.03 (1H, d, J=8Hz), 8.50-8.70 (1H, m)

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Analysis Calcd. for C₁₉H₂₄N₄O : C 70.34, H 7.46, N 17.27 Found : C 70.77, H 7.54, N 17.18

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A solution of 7-chloro-3-dimethylaminomethyl-2-methylimidazo[2,1-a]isoquinoline (3.4 g) in tetrahydrofuran (30 ml) was added dropwise to a solution of dimethyl sulfate (3.5 ml) in tetrahydrofuran (20 ml) with ice-cooling and stirring. After being stirred for 2 hours under the same conditions, the resulting precipitates were collected by filtration and washed with tetrahydrofuran and then ethanol to give 7-chloro-2-methyl-3-trimethylammoniomethylimidazo-[2,1-a]isoquinoline methylsulfate (4.4 g).

mp : 220 to 223°C (dec.)
IR (Nujol) : 1390, 1360, 1230, 1050, 750 cm⁻¹

Example 5

A solution of 3-dimethylaminomethyl-2methylimidazo[2,1-a]isoquinoline (7.7 g) in
tetrahydrofuran (20 ml) was added dropwise to a
solution of dimethyl sulfate (15 ml) in tetrahydrofuran
(40 ml) with ice-cooling and stirring, and the mixture
was stirred for 2 hours under the same conditions.
To the reaction mixture was added diethyl ether
(60 ml) and stirred with ice-cooling for 1 hour.
The resulting precipitates were collected by filtration,
washed with diethyl ether and recrystallized from
ethanol to give 2-methyl-3-trimethylammoniomethylimidazo[2,1-a]isoquinoline methylsulfate (10 g).

mp : 146 to 149°C

IR (Nujol) : 1630, 1550, 1510, 1240, 1220, 1010, 750 cm⁻¹

Example 6

Methyliodide (1.7 g) was added dropwise to a solution of 8-chloro-3-dimethylaminomethyl-2-methyl-imidazo[2,1-a]isoquinoline (2.8 g) in ethanol (34 ml)

at room temperature and the mixture was stirred for 14 hours. The resulting precipitate was collected by filtration, washed with ethanol and dried in a desiccator to give 8-chloro-2-methyl-3-trimethylammonio-methylimidazo[2,1-a]isoquinoline iodide (3.58 g).

mp : 169 to 170°C (dec.)

IR (Nujol) : 1630 cm⁻¹

NMR (DMSO-d₆, δ) : 2.57 (3H, s), 3.15 (9H, s),

5.03 (2H, s), 7.40 (1H, d, J=7.5Hz),

7.70 (1H, dd, J=9Hz, 2Hz), 8.05 (1H, d,

J=2Hz), 8.48 (1H, d, J=9Hz), 8.67 (1H, d,

Example 7

J=7.5Hz)

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The following compound was obtained according to a similar manner to that of Example 6.

7-Benzyloxy-2-methyl-3-trimethylammoniomethylimidazo-[2,1-a]isoquinoline iodide

mp : 193 to 195°C (dec.)

IR (Nujol) : 1600, 1560, 1510, 1270 cm⁻¹

NMR (DMSO-d₆, δ) : 2.57 (3H, s), 3.20 (9H, s),

5.1 (2H, s), 5.33 (2H, s), 7.13-7.77 (8H, m),

8.06 (1H, d, J=8Hz), 8.63 (1H, d, J=8Hz)

Example 8

A mixture of 7-chloro-2-methyl-3-trimethylammonio-methylimidazo[2,1-a]isoquinoline methylsulfate (2 g), sodium cyanide (0.32 g) in dimethyl sulfoxide (10 ml) was heated at 100°C with stirring for 3 hours and cooled to room temperature. The mixture was poured into water and extracted with chloroform. The extract was washed with brine, dried over magnesium sulfate and evaporated in vacuo. The residue was chromatographed on silica gel (20 g) with a mixture

of chloroform and methanol (10:1) as an eluent.

The eluates were evaporated in vacuo and the residual solid was recrystallized from ethanol to give 7-chloro-3-cyanomethyl-2-methylimidazo[2,l-a]-iscquinoline (0.5 g).

mp : 199 to 203°C

IR (Nujol): 2250, 1510, 1480, 1410, 1370 cm⁻¹

NMR (DMSO-d₆, δ): 2.45 (3H, s), 4.51 (2H, s),

7.37-7.85 (3H, m), 8.32-8.49 (2H, m)

Analysis Calcd. for C₁₄H₁₀ClN₃:

C 65.75, H 3.94, N 16.43

Found: C 66.12, H 4.27, N 16.55

Example 9

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The following compound was obtained according to a similar manner to that of Example 8.

3-Cyanomethyl-2-methylimidazo[2,1-a]isoquinoline mp : 167 to 170°C IR (Nujol) : 2250, 1390, 790, 710 cm⁻¹ NMR (CDCl₃, δ) : 2.43 (3H, s), 3.93 (2H, s), 7.03-7.80 (5H, m), 8.52-8.77 (1H, m) Analysis Calcd. for C₁₄H₁₁N₃ :

Found: C 76.17, H 5.06, N 19.07

C 75.99, H 5.01, N 18.99

Example 10

A mixture of 2-methyl-3-trimethylammoniomethyl-imidazo[2,1-a]isoquinoline methylsulfate (4.5 g) and imidazole (2.02 g) in ethanol (45 ml) was refluxed with stirring for 7 hours and evaporated in vacuo. To the residue was added water and the resulting precipitates were collected by filtration and recrystallized from a mixture of ethyl acetate and n-hexane to give 3-(1-imidazolylmethyl)-2-methylimidazo[2,1-a]isoquinoline (2.3 g).

mp: 189 to 191°C

IR (Nujol): 1640, 1610, 1570, 1550, 1510 cm⁻¹

NMR (CDCl₃, δ): 2.53 (3H, s), 5.36 (2H, s), 6.60-7.20 (3H, m), 7.23-7.80 (5H, m), 8.50-8.80 (1H, m)

Analysis Calcd. for C₁₆H₁₄N₄:

Analysis Calcd. for C₁₆H₁₄N₄: C 73.26, H 5.38, N 21.36 Found: C 72.77, H 5.37, N 21.00

10 Example 11

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A mixture of 2-methyl-3-trimethylammoniomethyl-imidazo[2,1-a]isoquinoline methylsulfate (3 g) and 2-methylimidazole (1.6 g) in ethanol (50 ml) was refluxed with stirring for 4 hours and evaporated in vacuo. To the residue was added water and the resulting precipitates were collected by filtration, dried and recrystallized from ethyl acetate to give 2-methyl-3-(2-methyl-1-imidazolylmethyl)imidazo-[2,1-a]isoquinoline (0.85 g).

20 mp: 182 to 184°C

IR (Nujol) : 3350, 3100, 1450, 1380, 1280 cm⁻¹

NMR (CDCl₃, δ) : 2.43 (3H, s), 2.50 (3H, s),

5.20 (2H, s), 6.58-7.83 (7H, m),

8.59-8.82 (1H, m)

Analysis Calcd. for $C_{17}^{H_{16}N_{4} \cdot 1/2H_{2}O}$: C 71.56, H 6.00, N 19.64, $H_{2}O$ 3.15 Found: C 71.43, H 5.91, N 19.37, $H_{2}O$ 3.30

Example 12

To a solution of 60% sodium hydride (0.235 g) in propargyl alcohol (9.36 ml) was added 7-chloro-2-methyl-3-trimethylammoniomethylimidazo[2,1-a]isoquinoline iodide (2.08 g) and the mixture was heated at 100°C with stirring for 2 hours. After being cooled, the mixture was evaporated in vacuo and the residue was dissolved

in chloroform. The solution was washed with water, dried over magnesium sulfate and evaporated in vacuo. The residue was purified by column chromatography on silica gel (20 g) with a mixture of chloroform and methanol (100:1) as an eluent to give 7-chloro-2-methyl-3-propargyloxymethylimidazo[2,1-a]isoquinoline. This compound was treated with a solution of hydrogen chloride in ethanol to give 7-chloro-2-methyl-3-propargyloxymethylimidazo[2,1-a]isoquinoline hydrochloride (1 g).

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mp: 210 to 216°C (dec.)

IR (Nujol): 3200, 2360, 1645, 1580 cm⁻¹

NMR (CF₃COOH, δ): 2.70 (lH, t, J=2Hz), 2.80

(3H, s), 4.47 (2H, d, J=2Hz), 5.30 (2H, s),

7.73-8.73 (5H, m)

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Example 13

The following compounds were obtained according to similar manners to those of Examples 8 and 10 to 12.

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(1) 7-Benzyloxy-3-cyanomethyl-2-methylimidazo[2,1-a]isoquinoline

mp : 189 to 191°C

IR (Nujol): 2250, 1600, 1570, 1550, 1260, 1010 cm⁻¹

NMR (CDCl₃, δ): 2.43 (3H, s), 3.83 (2H, s), 5.15 (2H, s), 7.00 (1H, d, J=8Hz), 7.20-7.77 (8H, m), 8.17 (1H, d, J=8Hz)

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8-Chloro-3-cyanomethyl-2-methylimidazo[2,l-a]isoquinoline

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8.03 (1H, d, J=2Hz), 8.27 (1H, d, J=7Hz),
                   8.42 (lH, d, J=9Hz)
            7-Chloro-3-methoxymethyl-2-methylimidazo[2,1-a]-
         (3)
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              isoquinoline
              mp : 148 to 149°C
              NMR (CDCl<sub>3</sub>, \delta): 2.53 (3H, s), 3.33 (3H, s),
                   4.75 (2H, s), 7.25-7.73 (3H, s), 8.02 (1H, d,
                   J=7.5Hz), 8.57 (1H, dd, J=7.5Hz, 2.5Hz)
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        (4) 3-Allyloxymethyl-7-chloro-2-methylimidazo[2,l-a]-
              isoquinoline hydrochloride
              mp: 173 to 175°C (dec.)
              IR (Nujol): 2360, 1650, 1585, 1522 cm<sup>-1</sup>
              NMR (D_2O, \delta): 2.64 (3H, s), 4.28 (2H, d, J=6Hz),
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                   4.90 (2H, s), 5.36 (1H, m), 5.60 (1H, m),
                   5.73-6.5 (lH, m), 7.30 (lH, d, J=8Hz),
                   7.15-7.70 (3H, m), 8.0 (1H, d, J=8Hz)
            7-Chloro-2-methyl-3-methylthiomethylimidazo[2,1-a]-
20
        (5)
              isoquinoline
             mp: 146 to 148°C
              IR (Nujol): 1560, 1500, 1360 cm<sup>-1</sup>
             NMR (CF_3COOH, \delta): 2.19 (3H, s), 2.69 (3H, s),
25
                   4.22 (2H, s), 7.87-8.72 (5H, m)
             3-(2-Aminoethyl) thiomethyl-7-chloro-2-methylimidazo-
        (6)
              [2,1-a]isoquinoline
             mp: 94°C
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             IR (Nujol): 3310, 1615, 1595, 1570, 1510 cm<sup>-1</sup>
             NMR (CDCl<sub>3</sub>, \delta): 1.83 (2H, s), 2.47 (3H, s),
                   2.20-2.97 (4H, m), 3.98 (2H, s), 7.20-7.70
                   (3H, m), 7.93 (1H, d, J=8Hz), 8.50 (1H, dd)
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J=2Hz, 7Hz)

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Ethanol (50 ml) was added to a mixture of 7-chloro-2-methyl-3-trimethylammoniomethylimidazo-[2,1-a]isoquinoline methylsulfate (2.3 g) and imidazole (0.95 g) in dimethyl sulfoxide (20 ml) and heated at 100°C with stirring for 5 hours. After the removal of ethanol by distillation, the residue was poured into water. The resulting precipitates were collected by filtration, dried and recrystallized from ethyl acetate. The obtained crystals were chromatographed on silica gel (16 g) with a mixture of chloroform and methanol (10:1) as an eluent. The eluates were evaporated in vacuo and the residual solid was recrystallized from ethyl acetate to give 7-chloro-3-(1-imidazolylmethyl)-2-methylimidazo[2,1-a]isoquinoline (0.7 g).

mp : 229 to 230°C

IR (Nujol): 1390, 1370, 1220, 1080, 790 cm⁻¹

NMR (CDCl₃, δ): 2.58 (3H, s), 5.42 (2H, s),

6.84-7.70 (7H, m), 8.48-8.63 (1H, m)

Analysis Calcd. for C16H13ClN4:

C 64.76, H 4.41, N 18.88

Found: C 64.95, H 4.58, N 19.02

The mother liquid was evaporated in vacuo and the residual solid was recrystallized from diethyl ether to give 7-chloro-3-ethoxymethyl-2-methylimidazo[2,l-a]isoquinoline (0.2 g).

mp: 94 to 96°C

IR (Nujol): 1380, 1360, 1180, 780 cm⁻¹

NMR (CDCl₃, δ): 1.21 (3H, t, J=7Hz), 2.50 (3H, s),

3.53 (2H, q, J=7Hz), 4,77 (2H, s),

7.23-8.61 (5H, m)

Analysis Calcd. for C15H15ClN20:

C 65.57, H 5.50, N 10.20, Cl 12.91

Found: C 65.66, H 5.42, N 10.37, Cl 13.21

A mixture of 3-cyanomethyl-2-methylimidazo[2,1-a]-isoquinoline (2.8 g), ammonium chloride (0.74 g) and sodium azide (0.9 g) in N,N-dimethylformamide (28 ml) was heated at 120-125°C with stirring for 18 hours. The solution was poured into water, acidified with concentrated hydrochloric acid and then neutralized with an aqueous solution of sodium bicarbonate. The resulting precipitates were collected by filtration, washed with water and recrystallized from a mixture of chloroform and methanol to give 2-methyl-3-[(1H-tetrazol-5-yl)methyl]imidazo[2,1-a]isoquinoline (2.9 g).

mp: 256 to 257°C

IR (Nujol): 3450, 2500 (broad), 1660, 1540 cm⁻¹

NMR (DMSO-d₆, δ): 2.12 (3H, s), 4.74 (2H, s),
6.70 (3H, br s), 7.18-8.55 (6H, m)

Analysis Calcd. for C₁₄H₁₂N₆·H₂O: C 59.56, H 4.99, N 29.77 Found: C 59.31, H 5.03, N 29.33

round . C 33.31, h 3.03,

Example 16

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A solution of 7-chloro-2-methylimidazo[2,1-a]isoquinoline (4 g) in N,N-dimethylformamide (40 ml) was
added dropwise to a mixture of N,N-dimethylformamide
(5.7 ml) and phosphoryl chloride (1.86 ml) under ice
cooling. After being stirred at 85-90°C for 6 hours,
the mixture was poured into ice-water, basified with
aqueous sodium hydroxide, heated at 80°C for 20
minutes and cooled. The resulting precipitate was
collected by filtration, washed with water and recrystallized from methanol to give 7-chloro-3-formyl-2methylimidazo[2,1-a]isoquinoline (0.356 g).

mp : 213 to 214°C IR (Nujol) : 1655 cm^{-1} 35NMR (CF₃COOH, δ) : -3.13 (3H, s), 8.10-8.87 (4H, m), 9.72 (1H, d, J=7.5Hz), 10.45 (1H, s)

Isoamyl nitrite (140 g) was added to a suspension of 7-chloro-2-methylimidazo[2,1-a]isoquinoline (17.1 g) in dioxane (170 ml) at 50°C and the mixture was refluxed for 15 minutes. After being cooled, the resulting precipitate was coilected by filtration, washed successively with dioxane and diethyl ether and dried in a desiccator to give 7-chloro-2-methyl-3-nitrosoimidazo-[2,1-a]isoquinoline (7.7 g).

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mp : >190°C (dec.)

IR (Nujol): 1620, 1585, 1350, 1250, 1200 cm⁻¹

Example 18

Zinc powder (10.2 g) was added portionwise to a mixture of 7-chloro-2-methyl-3-nitrosoimidazo[2,l-a]-isoquinoline (7.7 g) and acetic acid (100 ml) in water (77 ml) over a period of 7 hours and then the mixture was filtered by suction. The filtrate was evaporated in vacuo and the residue was extracted with chloroform after an addition of aqueous sodium bicarbonte to it. The extract was washed with water and evaporated in vacuo. The residual solid was washed with ethyl acetate and dried in a desiccator to give 3-amino-7-chloro-2-methylimidazo[2,l-a]isoquinoline (5.85 g).

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IR (Nujol) : 3250, 3150, 1610, 1590, 1500 cm⁻¹
NMR (DMSO-d₆, δ) : 2.33 (3H, s), 4.83 (2H, broad s),
7.28 (1H, d, J=8Hz), 7.43-7.70 (2H, m),
8.10 (1H, d, J=8Hz), 8.30 (1H, dd, J=4Hz,
7Hz)

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Example 19

Ethyl N-cyanoformimidate (3.2 g) was added to a suspension of 3-amino-7-chloro-2-methylimidazo[2,1-a]-isoquinoline (2.3 g) in ethanol (60 ml) and the mixture was stirred at room temperature for 45 hours.

The mixture was evaporated in vacuo and the residual solid was washed successively with diethyl ether and ethyl acetate to give N-cyano-N'-(7-chloro-2-methylimidazo-[2,1-a]isoquinolin-3-yl) formamidine (1.0 g).

mp : 203 to 206°C (dec.) IR (Nujol) : 3240, 2200, 1620, 1595 cm $^{-1}$ NMR (DMSO-d₆, δ) : 2.33 (3H, s), 7.3-8.9 (6H, m)

Example 20

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Isopropylamine (2 ml) was added to a suspension of N-cyano-N'-(7-chloro-2-methylimidazo[2,1-a]isoquinolin-3-yl) formamidine (0.95 g) in water (1.5 ml) and the mixture was stirred at room temperature for 30 minutes. The resulting precipitate was collected by filtration, washed with water and dissolved in ethanol. To the solution was added a saturated solution of hydrogen chloride in ethanol and the mixture was evaporated in vacuo. The residual solid was washed successively with a mixture of diethyl ether and ethanol and isopropanol and then dried in a desiccator to give N-isopropyl-N'-(7-chloro-2-methylimidazo[2,1-a]-isoquinolin-3-yl) formamidine dihydrochloride (0.73 g).

NMR (DMSO-d₆, δ): 1.23 (3H, d, J=7Hz), 1.34 (3H, d, J=7Hz), 2.5 (3H, s), 4.0 (1H, m), 7.5-9.1 (6H, m)

189 to 190°C

Example 21

To a mixture of 3-(2-aminoethyl) thiomethyl-7-chloro-2-methylimidazo[2,l-a]isoquinoline (1.5 g) and triethylamine (0.5 g) in ethyl alcohol (35 ml) was added portionwise 3-chloro-1,2-benzisothiazole-1,l-dioxide (1.0 g). After being stirred at ambient temperature for 1 hour, the resulting precipitate was collected by filtration and washed several times with

ethyl alcohol to give 3-[2-[(7-chloro-2-methylimidazo-[2,1-a]isoquinolin-3-yl)methylthio]ethylamino]-1,2-benzisothiazole-1,1-dioxide (2.2 g).

mp : 291 to 294°C (dec.)

IR (Nujol) : 3300, 161.5, 1580 cm⁻¹

NMR (DMSO-d₆, δ) : 2.43 (3H, s), 2.77 (2H, t,

J=7Hz), 3.63 (2H, m), 4.3 (2H, s), 7.27-8.50

(9H, m), 9.4 (lH, t, J=6Hz)

Example 22

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To a solution of 7-chloro-2-methyl-3-methylthio-methylimidazo[2,1-a]isoquinoline (2 g) in chloroform (20 ml) was added portionwise 70% m-chloroperbenzoic acid (1.78 g). After being stirred under ice cooling for 30 minutes, the mixture was washed successively with aqueous sodium bicarbonate, water and brine, dried over magnesium sulfate, treated with activated charcoal and evaporated in vacuo. The residue was purified by column chromatography on silica gel (25 g) with a mixture of chloroform and methanol (50:1) as an eluent to afford a solid. This solid was recrystallized from ethanol to give 7-chloro-2-methyl-3-methylsulfinyl-methylimidazo[2,1-a]isoquinoline (1.36 g).

mp: 205 to 207°C (dec.)
IR (Nujo1): 1595, 1560, 1500, 1365, 1055, 1025 cm⁻¹
NMR (DMSO-d₆, δ): 2.42 (3H, s), 2.58 (3H, s),
4.41 (1H, d, J=15Hz), 4.77 (1H, d, J=15Hz),
7.27-7.83 (3H, m), 8.30-8.62 (2H, m)

Example 23

To a solution of 7-chloro-2-methyl-3-methylthio-methylimidazo[2,1-a]isoquinoline (2 g) in chloroform (40 ml) was added 70% m-chloroperbenzoic acid (3.74 g) under ice-cooling and then the mixture was stirred for 1 hour at room temperature. The mixture was washed

successively with aqueous sodium carbonate and water, dried over magnesium sulfate, treated with activated charcoal and evaporated in vacuo. The residue was recrystallized from a mixture of chloroform and methanol to give 7-chloro-2-methyl-3-methylsulfonylmethylimidazo-[2,1-a]isoquinoline (1.17 g).

IR (Nujol): 1560, 1500, 1310, 1120 cm⁻¹

NMR (CF₃COOH, δ): 2.82 (3H, s), 3.47 (3H, s),

5.18 (2H, s), 7.75-8.70 (5H, m)

Example 24

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A mixture of 7-chloro-3-cyanomethyl-2-methylimidazo-[2,1-a]isoquinoline (6 g) in a saturated solution of hydrogen chloride in ethanol (80 ml) was refluxed for 5 hours and evaporated in vacuo. To the residue was added aqueous sodium bicarbonate and extracted with chloroform. The extract was washed with brine, dried over magnesium sulfate and evaporated in vacuo. The crystalline residue was recrystallized from ethyl acetate to give 7-chloro-3-ethoxycarbonylmethyl-2-methyl-imidazo[2,1-a]isoquinoline (6.1 g).

mp : 137 to 138°C IR (Nujol) : 1715, 1590, 1565, 1195 cm⁻¹ NMR (CF₃COOH, δ) : 1.42 (3H, t, J=7Hz), 2.68 (3H, s), 4.32 (2H, s), 4.43 (2H, q, J=7Hz), 7.72-8.60 (5H, m)

Example 25

A solution of sodium hydroxide (1.25 g) in water (5 ml) was added to a mixture of 7-chloro-3-ethoxy-carbonylmethyl-2-methylimidazo[2,1-a]isoquinoline (4.75 g) and methanol (48 ml) and the mixture was refluxed for 1.5 hours. After being cooled, the mixture was evaporated in vacuo and lN hydrochloric acid (31.4 ml) was added to the residue under ice-cooling.

The resulting precipitate was collected by filtration, washed with water, dried in a desiccator and washed with a mixture of ethanol and methanol to give 3-carboxymethyl-7-chloro-2-methylimidazo[2,l-a]isoquinoline (3.2 g).

mp : 275 to 276°C (dec.) IR (Nujol) : 1690, 1585, 1500 cm⁻¹ NMR (CF₃COOH, δ) : 2.68 (3H, s), 4.32 (2H, s), 7.70-8.57 (5H, m)

I0

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Example 26

The following compound was obtained according to a similar manner to that of Example 25.

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3-Carboxy-7-chloro-2-methylimidazo[2,1-a]isoquinoline mp: 228 to 230°C (dec.) IR (Nujol): 2450, 1680, 1495 cm $^{-1}$ NMR (CF₃COOH, δ): 2.07 (3H, s), 7.77-8.63 (4H, m), 9.47 (1H, d, J=7.5Hz)

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Example 27

A mixture of 3-carboxy-7-chloro-2-methylimidazo-[2,1-a]isoquinoline (2.5 g) and thionyl chloride (20 ml) was refluxed for 1 hour and evaporated in vacuo. The residual solid was washed with benzene and dried in a desiccator to give 7-chloro-3-chloroformyl-2-methylimidazo-[2,1-a]isoquinoline hydrochloride (2.75 g).

IR (Nujol) : 1790 cm⁻¹

30 Example 28

To 28% ammonium hydroxide (50 ml) was added 7-chloro-3-chloroformyl-2-methylimidazo[2,1-a]isoquinoline hydrochloride (1.3 g) and the mixture was stirred under ice-cooling for 40 minutes. The resulting precipitate was collected by filtration, washed with water and recrystallized from a mixture of chloroform and methanol to give 3-carbamoyl-7-chloro-2-methylimidazo[2,1-a]-isoquinoline (0.75 g).

mp: 289 to 290°C (dec.)

IR (Nujol): 3340, 3160, 1630, 1595 cm⁻¹

NMR (CF₃COOH, δ): 3.00 (3H, s), 7.33-7.73 (2H, m), 7.73-7.93 (4H, m), 9.20 (1H, d, J=8Hz)

Example 29

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Thionyl chloride (0.56 ml) was added dropwise to a suspension of 3-carboxymethyl-7-chloro-2-methylimidazo-[2,1-a]isoquinoline (0.855 g) in methylene chloride (9 ml) and the mixture was stirred for 2 hours at room temperature. The resulting precipitate was collected by filtration, washed with methylene chloride and then treated with 28% ammonium hydroxide (20 ml) for 1 hour. The resulting precipitate was collected by filtration, washed with water and recrystallized from ethanol to give 3-carbamoylmethyl-7-chloro-2-methylimidazo[2,1-a]-isoquinoline (0.75 g).

mp: 284 to 286°C

IR (Nujol): 3350, 3100, 1665, 1610, 1510 cm⁻¹

NMR (CF₃COOH, δ): 2.73 (3H, s), 4.3 (2H, s),

7.73-8.57 (5H, m)

Example 30

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The following compound was obtained according to a similar manner to that of Example 29.

2-(7-Chloro-2-methylimidazo[2,1-a]isoquinolin-3-y1)acetohydroxamic acid
mp: 249 to 254°C (dec.)

IR (Nujol): 3200, 1645, 1410 cm⁻¹

NMR (CF₃COOH, δ): 2.73 (3H, s), 4.33 (2H, s),

7.7-8.7 (5H, m)

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Lithium aluminum hydride (0.668 g) was added portionwise to a solution of 7-chloro-3-ethoxycarbonyl-2-methylimidazo[2,1-a]isoquinoline (5.1 g) in tetrahydrofuran (100 ml) with ice-cooling under nitrogen atmosphere and the mixture was stirred at room temperature for 1.5 hours. After an addition of ethyl acetate, the mixture was evaporated in vacuo and the residue was extracted with hot ethanol. The extract was treated with activated charcoal and evaporated in vacuo. The residue was recrystallized from 90% ethanol to give 7-chloro-3-hydroxymethyl-2-methylimidazo[2,1-a]-isoquinoline (1.94 g).

mp : 231 to 232°C IR (Nujol) : 3120, 1570, 1500 cm $^{-1}$ NMR (CF₃COOH, δ) : 2.75 (3H, s), 5.40 (2H, s), 7.87-8.70 (5H, m)

Example 32

Acetic anhydride (0.71 g) was added dropwise to a suspension of 7-chloro-3-hydroxymethyl-2-methylimidazo-[2,1-a]isoquinoline (1.43 g) in pyridine (14 ml) and the mixture was stirred at room temperature for 24 hours. The mixture was evaporated in vacuo and the residue was extracted with chloroform after an addition of water to it. The extract was washed with water, dried over magnesium sulfate, treated with activated charcoal and evaporated in vacuo. The crystalline residue was recrystallized from a mixture of ethyl acetate and diisopropyl ether to give 3-acetoxymethyl-7-chloro-2-methylimidazo[2,1-a]isoquinoline (0.6 g).

mp: 137 to 139°C

IR (Nujol): 1730, 1570, 1510 cm⁻¹

NMR (CDCl₃, δ): 2.07 (3H, s), 2.53 (3H, s),

5.40 (2H, s), 7.23-7.75 (3H, m), 8.00 (1H, d,

J=7Hz), 8.50 (1H, dd, J=6.5, 2.5Hz)

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A mixture of 1-amino-5-chloroisoquinoline (6.1 g), ethyl 2-acetyl-2-bromoacetate (10.75 g) and sodium bicarbonate (14.36 g) in ethanol (72 ml) was refluxed for 2 hours and then filtered by suction. The filtrate was evaporated in vacuo, and the residue was dissolved in a mixture of chloroform and methanol. The solution was treated with silica gel (10 g) and filtered by suction. The filtrate was evaporated in vacuo and the residue was recrystallized from ethanol to give 7-chloro-3-ethoxycarbonyl-2-methylimidazo[2,l-a]-isoquinoline (5.83 g).

mp : 140 to 141°C
IR (Nujol) : 1690, 1415, 1260, 1190, 1100 cm⁻¹
NMR (CDCl₃, δ) : 1.46 (3H, t, J=7Hz), 2.73 (3H, s),
 4.40 (2H, q, J=7Hz), 7.20-7.73 (3H, m),
 8.43 (1H, dd, J=7Hz, 2.5Hz), 8.93 (1H, d,
 J=7.5Hz)

20 Example 34

To a suspension of 1-amino-5-chloroisoquinoline (10 g) in chloroform (150 ml) was added dropwise 3-chloro-2,4-pentanedione (11.3 g). The mixture was refluxed for 6 hours and then allowed to stand at room temperature over night. The mixture was washed successively with aqueous sodium bicarbonate and water, dried over magnesium sulfate and evaporated in vacuo. The residue was purified by column chromatography on silica gel (200 g) with chloroform as an eluent to afford a solid. This solid was recrystallized from ethyl acetate to give 3-acetyl-7-chloro-2-methylimidazo-[2,1-a]isoquinoline (1.3 g).

IR (Nujol): 1630, 1500 cm⁻¹

NMR (CF₃COOH, δ): 2.97 (3H, s), 3.15 (3H, s), 7.82-8.72 (4H, m), 9.76 (1H, d, J=7.5Hz)

The following compound was obtained according to similar manners to those of Examples 33 and 34.

7-Chloro-2,3-dimethylimidazo[2,1-a]isoquinoline mp: 167 to 169°C

IR (Nujol): 1595, 1568, 1505 cm⁻¹

NMR (CF₃COOH, δ): 2.67 (6H, s), 7.7-8.6 (5H, m)

What we claim is :

A compound of the formula :

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$$R^3$$
 R^1

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wherein R¹ is lower alkyl, R³ is hydrogen, halogen or ar(lower)alkoxy, and

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R is lower alkanoyl, nitroso, amino, carboxy, protected carboxy, carbamoyl, hydroxycarbamoyl, haloformyl, aminomethyleneamino which may be substituted with cyano or lower alkyl, iminomethylamino which may be substituted with cyano or lower alkyl, or a group of the formula : $-A-R^2$

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in which A is lower alkylene and R² is di(lower)alkylamino,

cyano, lower alkoxy,

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N-containing heterocyclic group which may have suitable substituent(s), lower

alkynyloxy, lower alkenyloxy, lower alkylthio, amino-

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(lower) alkylthio, lower

alkylsulfinyl, lower alkylsulfonyl, carboxy, protected carboxy, carbamoyl,

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hydroxycarbamoyl, hydroxy, lower alkanoyloxy,

heterocyclicamino (lower) alkylthio having two oxo groups, hydrogen or a group of the formula:

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$$\stackrel{\oplus}{\longrightarrow} \underset{\mathbb{R}^5}{\overset{\mathbb{R}^4}{\longleftarrow}}$$

in which R⁴, R⁵ and R⁶ are each lower alkyl and X is an acid residue,

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and pharmaceutically acceptable salts thereof.

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2. A compound of claim 1, wherein R³ is hydrogen, halogen or phenyl(lower)alkoxy, and R is lower alkanoyl, nitroso, amino, carboxy, esterified carboxy, carbamoyl, hydroxycarbamoyl, haloformyl, cyanoaminomethyleneamino, lower alkylaminomethyleneamino, cyanoiminomethylamino, lower alkyliminomethylamino, or a group of the formula: -A-R²

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in which A is lower alkylene and

R² is di(lower)alkylamino, cyano,
lower alkoxy, saturated 3 to
8-membered heteromonocyclic group
containing 1 to 4 nitrogen atom(s)
which have hydroxy(lower)alkyl,

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unsaturated 3 to 8-membered heteromonocyclic group containing 1 to

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4 nitrogen atom(s) which may have lower alkyl, lower alkynyloxy, lower alkenyloxy, lower alkylthio,

amino(lower)alkylthio, lower

alkylsulfinyl, lower alkylsulfonyl, carboxy, esterified carboxy, carbamoyl.

hydroxycarbamoyl, hydroxy, lower alkanoyloxy, benzisothiazolylamino-(lower)alkylthio having two oxo groups, hydrogen or a group of the formula:

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in which R^4 , R^5 and R^6 are each lower alkyl,

X is an acid residue.

3. A compound of claim 2, wherein

R is lower alkanoyl, nitroso, amino, carboxy, lower alkoxycarbonyl, carbamoyl, hydroxycarbamoyl, haloformyl, cyanoaminomethyleneamino, lower alkylaminomethyleneamino, cyanoiminomethylamino, lower alkyliminomethylamino, or a group of the formula: -A-R²

in which A is lower alkylene and

R² is di(lower) alkylamino, cyano, lower alkoxy, piperazinyl having hydroxy(lower) alkyl, tetrazolyl, imidazolyl, imidazolyl having lower alkyl, lower alkynyloxy, lower alkenyloxy, lower alkylthio, amino(lower) alkylthio, lower alkylsulfinyl, lower alkylsulfonyl, carboxy, lower alkoxycarbonyl, carbamoyl, hydroxycarbamoyl, hydroxy, lower alkanoyloxy, (l,l-dioxo-l,2-benzisothiazolyl) amino(lower) alkylthio, hydrogen or a group of the formula:

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$$\stackrel{\oplus}{-N} \stackrel{R^4}{\underset{R^5}{<}}$$

in which R^4 , R^5 and R^6 are each lower alkyl,

X is residue of lower alkylsulfuric acid, or halogen.

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4. A compound of claim 3, wherein R³ is halogen, and R is a group of the formula:

-A-R²

in which A is lower alkylene and \mathbb{R}^2 is cyano.

5. A compound of claim 4, which is 7-chloro-3-cyanomethyl-2-methylimidazo[2,1-a]isoquinoline.

6. A process for preparing a compound of the formula:

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$$\mathbb{R}^3 \xrightarrow{\mathbb{N}} \mathbb{R}^1$$

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wherein R¹ is lower alkyl,

R³ is hydrogen, halogen or ar(lower)alkoxy,
and

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R is lower alkanoyl, nitroso, amino, carboxy, protected carboxy, carbamoyl, hydroxycarbamoyl, haloformyl, aminomethyleneamino which may be substituted with cyano or lower alkyl, iminomethylamino which may be substituted with cyano or lower alkyl, or

a group of the formula: $-A-R^2$ in which A is lower alkylene and R² is di(lower)alkylamino, cyano, lower alkoxy, N-containing heterocyclic group which may have suitable substituent(s), lower alkynyloxy, lower alkenyloxy, lower alkylthio, amino(lower)alkylthio, lower alkylsulfinyl, lower alkylsulfonyl, carboxy, protected carboxy, carbamoyl, hydroxycarbamoyl, hydroxy, lower alkanoyloxy, heterocyclicamino(lower)alkylthio having two oxo groups, hydrogen or a group of the formula :

$$\begin{array}{c}
X^{\bigcirc} \\
\stackrel{\longleftarrow}{-N} & \stackrel{\mathbb{R}^4}{=} & \text{in which } \mathbb{R}^4, \mathbb{R}^5 \text{ and} \\
\mathbb{R}^5 & \mathbb{R}^6 & \text{are each lower} \\
& & \text{alkyl and} \\
& & \text{X is an acid} \\
& & \text{residue,}
\end{array}$$

or a salt thereof, which comprises,

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(1) reacting a compound of the formula:

$$\mathbb{R}^3$$
 \mathbb{N} \mathbb{R}^1

wherein R^1 and R^3 are each as defined above, or a salt thereof with a compound of the formula:

wherein R_a² is di(lower)alkylamino or N-containing heterocyclic group which may have suitable substituent(s),

or a salt thereof and a compound of the formula:

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to give a compound of the formula :

wherein R^1 , R^3 and R_a^2 are each as defined above, or a salt thereof, or

(2) reacting a compound of the formula:

wherein R^1 , R^3 , R^4 , R^5 and A are each as defined above,

or a salt thereof with a compound of the formula:

wherein R^6 and X are each as defined above, to give a compound of the formula :

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wherein R^1 , R^3 , R^4 , R^5 , R^6 , A and X are each as defined above, or

(3) reacting a compound of the formula:

$$\begin{array}{c|c}
& \oplus & R^4 \\
& & \times & \times & R^5 \\
& & & \times & \times & R^6
\end{array}$$

wherein R¹, R³, R⁴, R⁵, R⁶, A and X are each as defined above,

with a compound of the formula: H-R²

wherein R² is cyano, lower alkoxy, di(lower)
alkylamino, N-containing heterocyclic group which may have suitable substituent(s), lower alkynyloxy, lower alkynyloxy, lower alkenyloxy, lower alkylthio, amino(lower)alkylthio or heterocyclicamino(lower)alkylthio having two oxo groups,

or a salt thereof to give a compound of the

or a salt thereof to give a compound of the formula :

wherein R^1 , R^3 , R_b^2 and A are each as defined above, or a salt thereof, or

(4) reacting a compound of the formula:

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wherein R¹, R³ and A are each as defined above, or a salt thereof with an azide compound to give a compound of the formula:

$$\mathbb{R}^{3} + \mathbb{I}^{N} = \mathbb{R}^{2}$$

wherein R^1 , R^3 and A are each as defined above, and $R_c^2 \text{ is tetrazol-5-yl,}$ or a salt thereof, or

(5) subjecting a compound of the formula:

$$\mathbb{R}^3$$

wherein R¹ and R³ are each as defined above, or a salt thereof to formylation reaction to give a compound of the formula:

wherein \mathbf{R}^1 and \mathbf{R}^3 are each as defined above, or a salt thereof, or

(6) subjecting a compound of the formula:

$$\mathbb{R}^3$$
 \mathbb{R}^1

wherein R^1 and R^3 are each as defined above, or a salt thereof to nitrosation reaction to give a compound of the formula:

wherein R^1 and R^3 are each as defined above, or a salt thereof, or

wherein R^1 and R^3 are each as defined above,

or a salt thereof to reduction reaction to give a compound of the formula:

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wherein R^1 and R^3 are each as defined above, or a salt thereof, or

(8) reacting a compound of the formula:

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wherein R¹ and R³ are each as defined above, or a salt thereof with a compound of the formula:

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wherein R⁷ is lower alkyl, to give a compound of the formula:

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wherein R^1 and R^3 are each as defined above, and R_a is cyanoaminomethyleneamino or cyanoiminomethylamino,

or a salt thereof, or

(9) reacting a compound of the formula:

$$\mathbb{R}^3 = \mathbb{N} \mathbb{R}^{R_a}$$

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wherein R^1 , R^3 and R_a are each as defined above,

or a salt thereof with a compound of the formula:

wherein \mathbb{R}^8 is lower alkyl, or a salt thereof to give a compound of the formula :

wherein R¹ and R³ are each as defined above,

R_b is lower alkylaminomethyleneamino or lower alkyliminomethylamino, or a salt thereof, or

(10) reacting a compound of the formula:

$$\mathbb{R}^{3} \xrightarrow{\mathbb{N}} \mathbb{R}^{1} \mathbb{R}^{1-NH_{2}}$$

wherein R^1 , R^3 and A are each as defined above, and A^1 is lower alkylene,

or a salt thereof with a compound of the formula:

$$y^1-R^9$$

wherein R⁹ is heterocyclic group having two oxo groups, and Y is halogen, or a salt thereof to give a compound of the

formula:

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wherein R^1 , R^3 , R^9 , A and A^1 are each as defined above,

or a salt thereof, or

(11) subjecting a compound of the formula:

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$$\mathbb{R}^{3} = \mathbb{R}^{2} \mathbb{R}^{1}$$

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wherein R¹, R³ and A are each as defined above,

 R_d^2 is lower alkylthio, or a salt thereof to oxidation reaction to give a compound of the formula:

wherein R^1 , R^3 and A are each as defined above, and R_e^2 is lower alkylsulfinyl or lower

alkylsulfonyl, or a salt thereof, or

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(12) reacting a compound of the formula:

wherein R^1 , R^3 and A are each as defined above, or a salt thereof with a compound of the formula:

wherein R¹⁰ is lower alkyl, and then subjecting the resultant compound to hydrolysis to give a compound of the formula:

wherein R^1 , R^3 , R^{10} and A are each as defined above,

or a salt thereof, or

(13) subjecting a compound of the formula:

wherein R^1 and R^3 are each as defined above, and

R_c is protected carboxy or protected carboxy(lower)alkyl,

or a salt thereof to elimination reaction of the carboxy protective group to give a compound of the formula;

wherein R^{1} and R^{3} are each as defined above, and

 $R_{ ext{d}}$ is carboxy or carboxy(lower)alkyl, or a salt thereof, or

(14) reacting a compound of the formula:

wherein R^1 , R^3 and $R_{\hat{d}}$ are each as defined above, or its reactive derivative at the carboxy group or a salt thereof with a compound of the formula:

wherein R^{ll} is hydrogen or hydroxy, or its reactive derivative at the amino group or a salt thereof to give a compound of the formula:

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wherein R^1 and R^3 are each as defined above,

R is carbamoyl, hydroxycarbamoyl, carbamoyl(lower)alkyl or 10 hydroxycarbamoyl (lower) alkyl, or a salt thereof, or

(15) subjecting a compound of the formula:

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$$\mathbb{R}^3 + \mathbb{R}_{\mathbb{R}^1}^{\mathbb{R}_{\mathbb{R}^1}}$$

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wherein R^1 and R^3 are each as defined above,

R_f is carboxy or esterified carboxy, or a salt thereof to reduction reaction to give a compound of the formula:

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wherein R¹ and R³ are each as defined above, or a salt thereof, or

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(16) reacting a compound of the formula:

wherein R^1 , R^3 and A are each as defined above, or a salt thereof with a compound of the formula :

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wherein R¹² is lower alkyl, or its reactive derivative at the carboxy group or a salt thereof to give a compound of the formula:

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wherein R^1 , R^3 , R^{12} and A are each as defined above, or a salt thereof, or

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(17) reacting a compound of the formula:

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wherein R³ is as defined above, or a salt thereof with a compound of the formula:

wherein R¹ and R are each as defined above,

and

Y² is halogen,

or a salt thereof to give a compound of the formula:

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wherein R, R^1 and R^3 are each as defined above, or a salt thereof.

7. A compound of the formula:

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$$R_a^3$$

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wherein R^1 is lower alkyl, and R^3 is halogen or ar(lower)alkoxy, and a salt thereof.

8. A process for preparing a compound of the formula:

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wherein R^1 is lower alkyl, and R^3 is halogen or ar(lower)alkoxy, or a salt thereof,

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which comprises reacting a compound of the formula :

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wherein R_a^3 is as defined above, or a salt thereof with a compound of the formula :

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wherein R^{1} is as defined above, and Y^{3} is halogen.

- 9. A pharmaceutical composition comprising an effective amount of a compound of claim 1 or pharmaceutically acceptable salt thereof in association with a pharmaceutically acceptable, substantially non-toxic carrier or excipient.
- 20 10. Use of the compound of claim 1 or pharmaceutically acceptable salt thereof for the manufacture of a medicament for treatment of ulcer in human being and animals.

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EUROPEAN PATENT APPLICATION

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[54] Imidazoisoquinoline compounds and processes for preparation thereof.

5 Imidazoisoquinoline compounds of the formula:

m wherein

R1 is lower alkyl,

R3 is hydrogen, halogen or ar(lower)alkoxy, and

R is lower alkanoyl, nitroso, amino, carboxy, protected carboxy, carbamoyl, hydroxycarbamoyl, haloformyl, aminomethyleneamino which may be substituted with cyano or lower alkyl, iminomethylamino which may be substituted with cyano or lower alkyl, or a group of the formula: -A-R² in which A is lower alkylene and

R² defines certain substituents,

pharmaceutical saits thereof and pharmaceutical compositions containing them as an active ingredient. The invention also relates to intermediates of formula

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and its preparation.



EUROPEAN SEARCH REPORT

EP 85 10 7218

ategory	Citation of document with indication, where appropriate, of relevant passages		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
Х	of relevant passages CHEMICAL ABSTRACTS volume 95, no. 5, 3rd August 1981, page 749, abstract no. 42984r; T.A. KUZMENKO et al.: "Studies on imidazo(2,1-a)isoquinoline derivatives. 2. Synthesis and reactions of esters of the imidazol(2,1-a)isoquinoline series"; & KHIM. GETEROTSIKL. SOEDIN 1980, (12), 1656-61		1-3	C 07 D 471/04 A 61 K 31/47 // (C 07 D 471/04 C 07 D 235:00 C 07 D 221:00)
X	CHEMICAL ABSTRACTS volume 94, no. 15, 13 pages 688,689, abstrav.V.V. KUZMENKO et al. "1-Chlorobenzotriazohetarylating agent"; GETEROTSIKL. SOEDIN	act no. 121408p; : le as a & KHIM.	1-3	
A	EP-A-0 068 378 (SCH * claims 1-3,5,11,13	ERING CORP.) *	1,6-10	TECHNICAL FIELDS
A	DE-A-2 551 868 (GRUPPO LEPETIT SPA) * examples 4-19,42-45 *		1,6-8	SEARCHED (Int. Cl.4) C 07 D 471/00
A	DE-A-2 523 876 (ASPRO-NICHOLAS LTD.) * examples 1,2 *		1,6-8	
A	CHEMICAL ABSTRACTS volume 99, no. 13, 26th September 1983, page 580, abstract no. 105177k; T.A. KUZMENKO et al.: "Studies of imidazo(2,1-a)isoquinoline derivatives. 3. Acylation of 2-substituted derivatives of imidazo(2,1-a)isoquinoline"; & KHIM. GETEROTSIKL. SOEDIN 1983, (6), 811-15			
	The present search report has b			
	Place of search	Date of completion of the se		Examiner
ĝ	BERLIN	07-12-1988	H	ASS C V F
Y Y A	CATEGORY OF CITED DOCUME : particularly relevant if taken alone : particularly relevant if combined with an document of the same category : technological background : non-written disclosure : intermediate document	E: earlier p after the nother D: documer L: documer	T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons &: member of the same patent family, corresponding document	